

Supplementary data for this paper are available from the IUCr electronic archives (Reference: BM1205). Services for accessing these data are described at the back of the journal.

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–S19.
- Casellato, U., Guerriero, P., Thamburini, S., Vigato, P. A. & Graziani, R. (1986). *Inorg. Chim. Acta*, **119**, 215–229.
- Fenton, D. E., Casellato, V., Vigato, P. A. & Vidali, M. (1982). *Inorg. Chim. Acta*, **62**, 57–66.
- Hodgkin, J. H. (1984). *Aust. J. Chem.* **37**, 2371–2374.
- Kurtz, D. M. (1990). *Chem. Rev.* **90**, 585–606.
- Nardelli, M. (1995). *J. Appl. Cryst.* **28**, 659.
- Sheldrick, G. M. (1990). *SHELXTL/PC User Manual*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Sheldrick, G. M. (1993). *SHELXL93. Program for the Refinement of Crystal Structures*. University of Göttingen, Germany.
- Siemens (1994). *XSCANS Users Manual*. Version 2.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.

Acta Cryst. (1998). **C54**, 367–368

8-Acetyl-4-methyl-9-phenylthio-7,8,9,10-tetrahydro-7,8-benzocoumarin†

KANDASAMY CHINNAKALI,^{a‡} HOONG-KUN FUN,^a KAMARAJ SRIRAGHAVAN^b AND VAYLAKKAVOOR T. RAMAKRISHNAN^b

^aX-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia, and ^bDepartment of Organic Chemistry, University of Madras, Guindy Campus, Chennai 600 025, India. E-mail: hkfun@usm.my

(Received 27 May 1997; accepted 23 October 1997)

Abstract

The coumarin and phenyl rings of the title molecule, C₂₂H₂₀O₄S, are individually planar. The tetrahydrobenzene ring adopts a half-chair conformation. The crystal structure is stabilized by C—H···O hydrogen bonds involving the carbonyl O atoms.

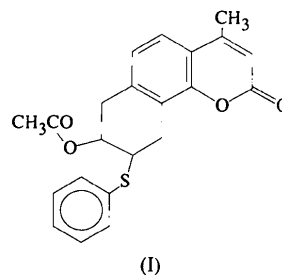
Comment

The coumarin sub-unit is of interest because it is found in many natural products displaying diverse biological activities. The range of compounds includes

† Alternative name: 4-methyl-9-phenylthio-7,8,9,10-tetrahydrobenzo[*h*]coumarin-8-yl acetate.

‡ On leave from: Department of Physics, Anna University, Chennai 600 025, India.

antifungal compounds, anticoagulants, and compounds active against psoriasis and carcinogens (Parrish *et al.*, 1974; Barry & Toste, 1996). The amino- and hydroxycoumarin derivatives are widely used in dye lasers (Maeda, 1984). The crystal structure determination of the title compound, (I), was undertaken as part of our structural studies on coumarin derivatives.



Bond lengths and valence angles in the benzocoumarin ring system are comparable with those observed in related derivatives (Chinnakali *et al.*, 1998; Kumar *et al.*, 1997). The coumarin moiety is planar with a maximum deviation of 0.030 (2) Å for C6. The tetrahydrobenzene ring adopts a half-chair conformation with asymmetry parameter $\Delta C_2(C7-C8) = 0.013$ (1) (Nardelli, 1983a). The thiophenyl and acetyl groups are planar and make dihedral angles of 62.08 (5) and 95.84 (7)°, respectively, with the best plane through atoms C7, C8, C12 and C15 of the tetrahydrobenzene ring. The carbonyl oxygen is involved in C—H···O hydrogen bonds, the geometries of which are given in Table 2.

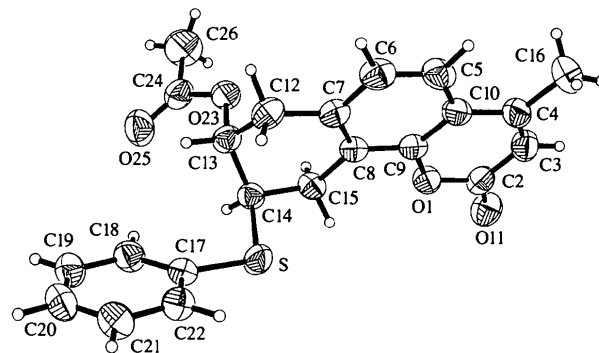


Fig. 1. The structure of the title compound showing 50% probability displacement ellipsoids and the atom-numbering scheme.

Experimental

Ring opening of the compound 4-methyl-7,10-dihydro-7,8-benzocoumarin-8,9-oxide with thiophenyl furnished two regioisomeric hydroxycoumarins, which on acetylation gave the corresponding acetylated compounds (Sriraghavan, 1998). Single crystals were grown by slow evaporation of the compound from a chloroform–methanol solution.

Crystal dataC₂₂H₂₀O₄S $M_r = 380.44$

Monoclinic

 $P2_1/c$ $a = 9.3028 (10) \text{ \AA}$ $b = 8.5506 (6) \text{ \AA}$ $c = 23.960 (2) \text{ \AA}$ $\beta = 97.636 (8)^\circ$ $V = 1889.0 (3) \text{ \AA}^3$ $Z = 4$ $D_x = 1.338 \text{ Mg m}^{-3}$ D_m not measured**Data collection**

Siemens P4 diffractometer

 $\theta/2\theta$ scans

Absorption correction: none

5750 measured reflections

4303 independent reflections

3450 reflections with

 $I > 2\sigma(I)$ $R_{\text{int}} = 0.017$ **Refinement**Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.040$ $wR(F^2) = 0.116$ $S = 1.048$

4303 reflections

324 parameters

All H atoms refined

 $w = 1/[\sigma^2(F_o^2) + (0.0656P)^2 + 0.183P]$ where $P = (F_o^2 + 2F_c^2)/3$ Mo $K\alpha$ radiation $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 27

reflections

 $\theta = 5.15\text{--}12.53^\circ$ $\mu = 0.196 \text{ mm}^{-1}$ $T = 293 (2) \text{ K}$

Rectangular block

 $0.84 \times 0.68 \times 0.54 \text{ mm}$

Colourless

 $\theta_{\text{max}} = 27.49^\circ$ $h = -1 \rightarrow 12$ $k = -1 \rightarrow 11$ $l = -31 \rightarrow 31$

3 standard reflections

every 97 reflections

intensity decay: <3%

 $(\Delta/\sigma)_{\text{max}} = -0.007$ $\Delta\rho_{\text{max}} = 0.217 \text{ e \AA}^{-3}$ $\Delta\rho_{\text{min}} = -0.238 \text{ e \AA}^{-3}$

Extinction correction: none

Scattering factors from

International Tables for Crystallography (Vol. C)

Supplementary data for this paper are available from the IUCr electronic archives (Reference: CF1196). Services for accessing these data are described at the back of the journal.

References

- Barry, M. T. & Toste, F. D. (1996). *J. Am. Chem. Soc.* **118**, 6305–6306.
- Chinnakali, K., Fun, H.-K., Sriraghavan, K. & Ramakrishnan, V. T. (1998). *Acta Cryst.* **C54**. In the press.
- Kumar, S., Chinnakali, K., Sivakumar, K., Fun, H.-K. & Sriraghavan, K. (1997). *Acta Cryst.* **C53**, 1854–1855.
- Maeda, M. (1984). In *Laser Dyes: Properties of Organic Compounds for Dye Lasers*. New York: Academic Press.
- Nardelli, M. (1983a). *Acta Cryst.* **C39**, 1141–1142.
- Nardelli, M. (1983b). *Comput. Chem.* **7**, 95–98.
- Parrish, J. A., Fitzpatrick, T. B., Tanenbaum, L. & Pathak, M. A. (1974). *New Engl. J. Med.* **291**, 206–209.
- Sheldrick, G. M. (1990). *SHELXTL/PC Users Manual*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Sheldrick, G. M. (1993). *SHELXL93. Program for the Refinement of Crystal Structures*. University of Göttingen, Germany.
- Siemens (1994). *XSCANS Users Manual*. Version 2.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Sriraghavan, K. (1998). PhD thesis, University of Madras, India. In preparation.

Acta Cryst. (1998). **C54**, 368–370

Diethyl 1-(3,4-Dichlorophenyl)-5-oxo-3-(2-thienyl)-2,2-pyrrolidinedicarboxylate

JAYANTA KUMAR RAY,^a ARINDAM CHAKRABORTY,^a
SUIT DAS ADHIKARI,^a KANDASAMY CHINNAKALI^{b†}
AND HOONG-KUN FUN^b

^aDepartment of Chemistry, Indian Institute of Technology, Kharagpur 721 302, India, and ^bX-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia. E-mail: hkfun@usm.my

(Received 24 April 1997; accepted 13 November 1997)

Abstract

In the title molecule, C₂₀H₁₉Cl₂NO₅S, the pyrrolidine ring is in an envelope conformation. The dichlorophenyl and thiophene rings are planar. Of the two ethoxy-carbonyl side chains, one is nearly planar but the other is distorted from planarity. The structure is stabilized by weak C—H···O hydrogen bonds and van der Waals interactions.

† On leave from Department of Physics, Anna University, Chennai 600 025, India.

Table 1. Selected torsion angles (°)

C12—C7—C8—C15	4.0 (2)	C12—C13—C14—C15	-57.7 (2)
C8—C7—C12—C13	-19.1 (2)	C7—C8—C15—C14	-15.7 (2)
C7—C12—C13—C14	46.0 (2)	C13—C14—C15—C8	41.5 (2)

Table 2. Hydrogen-bonding geometry (Å, °)

D—H···A	D—H	H···A	D···A	D—H···A
C6—H6···O11 ⁱ	1.01 (2)	2.58 (2)	3.470 (2)	147 (2)
C19—H19···O11 ⁱⁱ	0.98 (2)	2.45 (2)	3.378 (2)	159 (2)

Symmetry codes: (i) $1+x, y, z$; (ii) $-x, \frac{1}{2}+y, \frac{1}{2}-z$.

Data collection: XSCANS (Siemens, 1994). Cell refinement: XSCANS. Data reduction: XSCANS. Program(s) used to solve structure: SHELXTL/PC (Sheldrick, 1990). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993). Molecular graphics: SHELXTL/PC. Geometrical calculations PARST (Nardelli, 1983b).

The authors would like to thank the Malaysian Government and Universiti Sains Malaysia for research grant R&D No. 190-9609-2801. KC thanks the Universiti Sains Malaysia for a Visiting Post Doctoral Fellowship.